



## Invited paper

Emissions of nitrogen oxides and volatile organic compounds from liquefied petroleum gas-fueled taxis under idle and cruising modes<sup>☆</sup>

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## ABSTRACT

Liquefied petroleum gas (LPG) as an alternative fuel is increasingly used in mainland China, few reports are however available about emissions from LPG-fueled vehicles. In this study, 26 LPG-fueled taxis in Guangzhou, south China were tested using a chassis dynamometer to obtain their emission factors of nitrogen oxides (NO<sub>x</sub>) and volatile organic compounds (VOCs) under idle and cruising (10–60 km h<sup>-1</sup>) modes. The emission factors of NO<sub>x</sub> on average increased with speed from 4.13 g kg-fuel<sup>-1</sup> at idling to 71.1 g kg-fuel<sup>-1</sup> at 60 km h<sup>-1</sup> at a slope of 10.6 g kg-fuel<sup>-1</sup> per 10 km h<sup>-1</sup> increase in speed. Alkanes were the most abundant (71.9%) among the VOCs in the exhaust, followed by alkenes (25.2%), ethyne (2.7%), and aromatic species (0.2%). Emission factors of VOCs at idling averaged 8.24 g kg-fuel<sup>-1</sup>, higher than that of 6.23–7.36 g kg-fuel<sup>-1</sup> when cruising at 10–60 km h<sup>-1</sup>, but their ozone formation potentials (OFPs) were lower at idling (15.8 g kg-fuel<sup>-1</sup>) than under cruising (19.1–23.8 g kg-fuel<sup>-1</sup>) largely due to higher emission of more reactive alkenes under cruising mode. Emissions of both NO<sub>x</sub> and VOCs increased significantly with mileages. Measured emission factors of NO<sub>x</sub> and reactive VOCs in this study suggested that replacing the gasoline-powered taxis with the LPG-fueled taxis with LPG-gasoline bi-fuel engines and no efficient after-treatment devices would not benefit in reducing the emissions of ozone precursors, and strengthening the emission control for LPG vehicles with dedicated LPG engines and after-treatment converters, as did in Hong Kong, could further benefit in reducing the emission of photochemically active species when using LPG as alternative fuels.

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## 1. Introduction

In urban areas, vehicular emission contributes substantially to ambient nitrogen oxides (NO<sub>x</sub>) (Song and Xie, 2006; Mishra et al., 2016; He et al., 2016), volatile organic compounds (VOCs) (Liu et al., 2005; Song and Xie, 2006; Baker et al., 2008; Ho et al., 2009), and particulate matters (PM) (Eiguren-Fernandez et al.,

2004; Yu et al., 2013; Liu et al., 2014a; Worton et al., 2014). These primarily emitted pollutants, like NO<sub>x</sub> and VOCs, can also react through a series of complex chemical processes in the atmosphere to form secondary ozone (O<sub>3</sub>) (National Research Council, 1991) or secondary organic aerosols and nitrate (Gentner et al., 2017). Due to the crucial roles played by vehicular emission in air quality deterioration in megacities (Kirchstetter et al., 1996a, 1996b; Namdeo and Bell, 2005; Chan and Yao, 2008), sustained efforts have been made worldwide to reduce vehicular emissions, particularly in urban areas. Using cleaner fuels is one of the most important ways to lower vehicular emission, and switch from gasoline/diesel oils to liquefied petroleum gas (LPG) for buses and taxis in the public transportation sector is a widely adopted approach (Karamangil,

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2007; Myung et al., 2009; Adam et al., 2011; Guo et al., 2011; WLPGA and AEGPL, 2018).

As LPG is mostly composed of less reactive propane and butanes, previous studies have demonstrated that this fuel transition from gasoline/diesel oils to LPG could efficiently reduce emissions of reactive VOCs and particulate matters (Ristovski et al., 2005; Myung et al., 2009, 2012; 2014; Jang et al., 2014). Ristovski et al. (2005) compared the particle emissions between the unleaded petroleum vehicles and LPG vehicles (LPGVs) under different speeds and found that LPGVs were 'cleaner' with 70% fewer particle emissions. Myung et al. (2009) showed that LPGVs had significantly lower emission factors of particulate matters than diesel vehicles without a particulate filter. Myung et al. (2012) pointed out that LPGVs greatly excelled gasoline vehicles with over an order of magnitude lower emission factors of total hydrocarbons, benzene, and PM. Myung et al. (2014) showed that emissions of particle-bound polycyclic aromatic hydrocarbons (PAHs) from LPGVs decreased by over 90% when compared to those from gasoline vehicles. In Hong Kong, remote sensing and on-board measurements confirmed that LPG taxis in contrast to their diesel counterparts could greatly alleviate urban air pollution under real driving conditions (Ning and Chan, 2007; Lau et al., 2011).

While it is quite certain that the transition from gasoline/diesel to LPG would greatly reduce vehicular emissions of reactive VOCs and PM, it might not be so favorable in reducing vehicular emission of NO<sub>x</sub>. Early works revealed less NO<sub>x</sub> emitted from LPGVs when compared to that from gasoline vehicles (Pulkrabek, 1997), and much lower NO<sub>x</sub> emission from LPGVs than from diesel vehicles (Ning and Chan, 2007). Jang et al. (2014) however found LPGVs had slightly higher emissions of NO<sub>x</sub> than gasoline vehicles, and Zhang et al. (2013a) demonstrated, by conducting IG195 test for LPG taxis and on-road portable emission measurement system (PEMS) for LPG buses, that heavy-duty LPG buses did not show lower NO<sub>x</sub> emissions than diesel buses, and that replacing gasoline/diesel vehicle with LPGVs having no after-treatment equipment would not help mitigate NO<sub>x</sub> emissions as expected under real-world driving condition. Some studies further investigated the relationships between NO<sub>x</sub> emissions and driving speeds, yet inconsistent results came out possibly due to factors like mileages and performance of catalytic converters (Ning and Chan, 2007; Lau et al., 2011; Chikhi et al., 2014).

In the developing world, with the rapid increase in vehicle population in recent decades, traffic-related air pollution has become a serious environmental health issue, particularly in the urban street microenvironments. For example, in Guangzhou, a megacity in south China's Pearl River Delta region, traffic-related PM, CO and NO<sub>x</sub> frequently exceeded the guideline levels of Chinese National Air Quality Standard (Zhang and Mao, 1999; Wei et al., 1999; Fu et al., 2001; Qian et al., 2001) in the late 1990s when vehicles were fueled with gasoline and diesel oils: private vehicles and taxis were then all fueled with gasoline while public buses were fueled with diesel (Chan et al., 2003). High levels of hazardous air pollutants like particle-bound polycyclic aromatic hydrocarbons (PAHs) (Cheng et al., 1996; Bi et al., 2002) and aromatic hydrocarbons (Wang et al., 2002; Chan et al., 2003) in street canyons might pose health risks to commuters (Chan et al., 2002) and pedestrians (Zhao et al., 2004). Facing the deteriorating air quality, Guangzhou pioneered in mainland China to start converting all taxis and public buses to gasoline-LPG bi-fuel vehicles since 2003, following the practice in Hong Kong where dedicated LPGVs were introduced to replace diesel taxis since 1999 (Leung, 2011). Till 2014, the number of LPG-fueled taxis and buses in Guangzhou had reached 34.63 thousand (GZSB, 2015). Although these LPGVs only had a share of 1.38% in the total vehicle population, they accounted for 27% in the on-road vehicle fleet (Zhang et al., 2018a),

47% of the fleet-average vehicle kilometers traveled (VKT) (Zhang et al., 2013a) and consumed 9.2% of total fuel use in Guangzhou (GZSB, 2015). As for VOCs, this fuel conversion in Guangzhou has resulted in the decline of some hazardous air pollutants, like benzene, in ambient air; yet propane and butanes are becoming more and more abundant (Tang et al., 2007, 2008; Zhang et al., 2018a). Source apportionment results based on receptor models also revealed an increasingly important contribution of LPG to ambient VOCs in Guangzhou, such as 8–16% of ambient VOCs shared by LPG in 2004 (Liu et al., 2008) and 25.6% in 2008 (Zhang et al., 2012).

Very recently with the steady decline of ambient PM<sub>2.5</sub> in China's megacities, surface ozone instead has become an increasingly important air quality problem (Liu et al., 2018). For the control of ozone precursors (NO<sub>x</sub> and VOCs), the use of LPG as an alternative fuel for vehicles is becoming a debatable issue. LPGVs may have higher NO<sub>x</sub> emissions than conventional vehicles (Murillo et al., 2005; Yang et al., 2007; Zhang et al., 2013a). Meanwhile, results from a tunnel study in Guangzhou showed that LPG-related hydrocarbons, including propane, n-butane, and i-butane, already shared 59% of VOCs in the tunnel exhaust with estimated emission factors of propane over 30 times higher than that in Hong Kong (Zhang et al., 2018a). Although propane and butanes are not so photochemically reactive, high levels of these species accumulated in ambient air may also contribute substantially to ground-level O<sub>3</sub> formation (Blake and Rowland, 1995; Farmer et al., 2011; Lyu et al., 2016).

For policy-makers, extensive emission characterization for LPGVs is a prerequisite for taking any further regulatory or control steps. Characterizing emissions of NO<sub>x</sub> and VOCs from LPGVs, however, is still far from enough. Schifter et al. (2000) obtained average emission factors of NO<sub>x</sub> and hydrocarbons from retrofitted LPGVs under the Mexican official DGN-AA-II-1980 cycle. Chang et al. (2001) measured emission factors of 56 VOCs under the US FTP-75 cycle for vehicles using liquefied petroleum gas as an alternative fuel in Taipei. Guo et al. (2011) conducted chassis dynamometer tests at idling and 4 constant speeds to gain emission factors of individual VOC species from LPG taxis in Hong Kong. Myung et al. (2014) measured the emission factors of BTEX for the LPG-dedicated vehicles in Korea and found they varied greatly under NIER modes. While LPG consumption as alternative fuels is increasing in mainland China, very few reports are available for characterizing emissions from LPGVs except that a few studies reported compositions of VOCs released from LPGVs exhaust at idling (Lu et al., 2003; Wang et al., 2006; Lai et al., 2009).

In this study, emissions of NO<sub>x</sub> and VOCs from LPG-fueled vehicles in Guangzhou were tested at idling and different driving speeds using a chassis dynamometer. The study aims to obtain emission factors of NO<sub>x</sub> and VOCs for LPG taxis that are needed for assessing the potential impact of LPGV emissions on air quality and to gain source profiles of VOCs from LPGVs, and to explore the influence of speed and mileage on emissions from LPGVs.

## 2. Methodology

### 2.1. Descriptions of tested LPGVs

In this study, 26 LPG-fueled taxis were tested on a chassis dynamometer to attain the emission factors of CO<sub>2</sub>, CO, NO<sub>x</sub> (NO+NO<sub>2</sub>), and VOCs. These taxis were randomly recruited from the in-use taxis fleet in Guangzhou. The taxis fleet in Guangzhou began to switch from gasoline fueling to LPG fueling in 2003 by retrofitting original gasoline engines to LPG-gasoline bi-fuel engines, and the retrofit for running taxis in Guangzhou was completed by the end of 2006. According to Guangzhou Municipal

Ecological Environment Bureau, annual VKT by taxis was approximately  $14 \times 10^4 \text{ km veh}^{-1} \text{ yr}^{-1}$ . As taxis are mandatorily scrapped after running at most 8 years, mileages of running taxis typically span  $(0\text{--}100) \times 10^4 \text{ km}$  in Guangzhou. The tested vehicles in this study are all LPG-powered taxis with retrofitted bi-fuel engines. They were 1–5 years in vehicle age with mileages ranging  $(11.0\text{--}77.9) \times 10^4 \text{ km}$ , largely representing the real on-road fleet of in-use LPG-fueled taxis in Guangzhou. The specific data for the selected taxis are listed in Table S1.

## 2.2. Chassis dynamometer testing system

The selected LPG-fueled taxis were tested using a chassis dynamometer (Foshan Analytical Instrument Co., Ltd, China) inside the campus of the Guangzhou Institute of Geochemistry, Chinese Academy of Sciences (GIG-CAS). In this study, exhausts from the taxis were tested at hot idling and six constant speeds of  $10 \text{ km h}^{-1}$ ,  $20 \text{ km h}^{-1}$ ,  $30 \text{ km h}^{-1}$ ,  $40 \text{ km h}^{-1}$ ,  $50 \text{ km h}^{-1}$ , and  $60 \text{ km h}^{-1}$ , respectively. The tested taxis were operated by an experienced driver from the local motor vehicle testing center. During a test, the chassis dynamometer control platform gives an accurate reading of parameters including speed and torque. The exhaust from a testing taxi was sampled into a 4L Teflon sampling bag (Delin Gas Packing Co., Ltd, Dalian, China) by an Apex sampling pump (CASELLA CEL, UK) at a flow rate of  $0.5 \text{ L min}^{-1}$  for 6 min. The air samples were used for the analysis of VOCs, CO, and  $\text{CO}_2$ . Trace gases including  $\text{NO}_x$  ( $\text{NO} + \text{NO}_2$ ) in the exhaust were also directly measured by a flue gas analyzer (F-550, WOHLER, Germany) (Wu et al., 2020).

## 2.3. Air sample analysis

After statically diluted 1000 times by pure nitrogen, the air samples were analyzed for mixing ratios of VOCs with an automated Preconcentrator (Model 7100, Entech Instruments Inc., CA, USA) coupled with a gas chromatography (Model 6890)-mass selective detector (Model 5973N)/flame ionization detector/electron capture detector (GC-MSD/FID/ECD) (Agilent Technologies, CA, USA) (Wang and Wu, 2008; Zhang et al., 2012, 2013b). VOCs in the samples were identified by their retention times and mass spectra, and were quantified by an external calibration method (Zhang et al., 2012). Method detection limits for the VOCs were all below 50 pptv. CO concentrations were measured using a gas chromatograph (Model 6980, Agilent Technologies, CA, USA) with a flame ionization detector and a packed column (5A molecular sieve, 60/80mesh,  $3\text{m} \times 1/8 \text{ in.}$ ) (Zhang et al., 2012).  $\text{CO}_2$  was analyzed with an HP 4890D gas chromatograph (Zhang et al., 2018a).

## 2.4. Quality assurance and quality control

Before sampling the exhaust from the LPG taxis, each sampling bag was cleaned by filling and evacuating pure nitrogen gas repeatedly at least five cycles. To check if the sampling bags were clean, the bags were again filled with humidified pure nitrogen at the end of the filling/evacuating cycles and stored in the laboratory for at least 24h, and then analyzed in the same way as samples. If all the target compounds were not found or were below the method detection limits, the bags were certificated as clean.

## 2.5. Emission factor

The emission factors for a species  $p$  emitted from taxis are calculated using the carbon mass balance method (Kirchstetter et al., 1996a, 1996b; Kean et al., 2000; Martins et al., 2006; Liu et al., 2014b; Deng et al., 2020; Zhang et al., 2020) as:

$$E_p = \frac{\Delta(P)}{\Delta(\text{CO}_2) + \Delta(\text{CO}) + \Delta(\text{VOCs})} \times W_C \quad (1)$$

Where  $E_p$  is the emission factor for species  $p$ ,  $W_C = 0.821$  is the weight fraction of carbon in LPG with a propane/butanes ratio of 70:30 (Wang, 2010);  $\Delta(P)$  is the background-corrected concentration of pollutant  $p$ , and  $\Delta(\text{CO}_2)$ ,  $\Delta(\text{CO})$ , and  $\Delta(\text{VOCs})$  are the background-corrected carbon concentrations for  $\text{CO}_2$ , CO and VOCs, respectively. For the convenience of comparison with other studies, the fuel-based emission factors in  $\text{g kg-fuel}^{-1}$  can be converted to mileage-based emission factors in  $\text{g km}^{-1}$ :

$$E_{p\text{-mileage}} = E_{p\text{-fuel}} \times G$$

Where  $G$  is the LPG consumption in kg per 100 km, and here we assume  $G = 6.88 \text{ kg}/100 \text{ km}$  ( $13 \text{ L}/100 \text{ km}$ ) according to Guangzhou Municipal Ecological Environment Bureau.

## 3. Results and discussion

### 3.1. Emission factors of $\text{NO}_x$ ( $E_{\text{NO}_x}$ )

As showed in Table 1, the average  $E_{\text{NO}_x}$  at idling and at constant speeds of 10, 20, 30, 40, 50 and  $60 \text{ km h}^{-1}$  were  $4.13 \pm 4.21$ ,  $19.8 \pm 28.5$ ,  $31.9 \pm 51.2$ ,  $34.0 \pm 45.9$ ,  $44.4 \pm 53.1$ ,  $61.1 \pm 58.5$ ,  $71.1 \pm 60.4 \text{ g kg-fuel}^{-1}$ , respectively. Largely due to the lack of after-treatment system or the inefficiency of after-treatment system (He et al., 2019), the tested taxis in this study had the highest  $E_{\text{NO}_x}$  when compared to LPGVs in other studies (Table S2). Chikhi et al. (2014) found the average  $E_{\text{NO}_x}$  of gasoline-LPG vehicles without catalyzers was 4–5 times after treatment. Lyu et al. (2016) also reported that a catalytic converter replacing the project implemented in Hong Kong dramatically reduced NO emissions from LPG-fueled vehicles by  $88.6 \pm 0.7\%$ .

Fig. 1 shows the changes in  $E_{\text{NO}_x}$  with driving speeds. The average  $E_{\text{NO}_x}$  rose steadily from idling to  $60 \text{ km h}^{-1}$  with an increase of  $\sim 10.6 \text{ g kg-fuel}^{-1}$  per  $10 \text{ km h}^{-1}$  increase in speed. Chikhi et al. (2014) also observed a similar pattern of  $E_{\text{NO}_x}$  versus speed when testing LPGVs without after-treatment devices. However, Lau et al. (2011) revealed a decreasing trend of  $E_{\text{NO}_x}$  with speed for LPGVs with catalytic converters in Hong Kong. Even some LPG-taxis in Guangzhou did equip with catalytic converters, illegal tampering, and malfunction or deterioration of catalytic converters might also result in increasing  $E_{\text{NO}_x}$  with speed, just like vehicles without after-treatment devices (Lau et al., 2011; Chikhi et al., 2014; He et al., 2019).

### 3.2. Emission factors of VOCs ( $E_{\text{VOCs}}$ )

#### 3.2.1. Emission factors

As shown in Table 1 and Fig. 2, on average the total emission factor of VOCs for the tested LPG taxis at idling was  $8.24 \text{ g kg-fuel}^{-1}$ , higher than those ranging  $6.23\text{--}7.36 \text{ g kg-fuel}^{-1}$  when driving at constant speeds from 10 to  $60 \text{ km h}^{-1}$ .

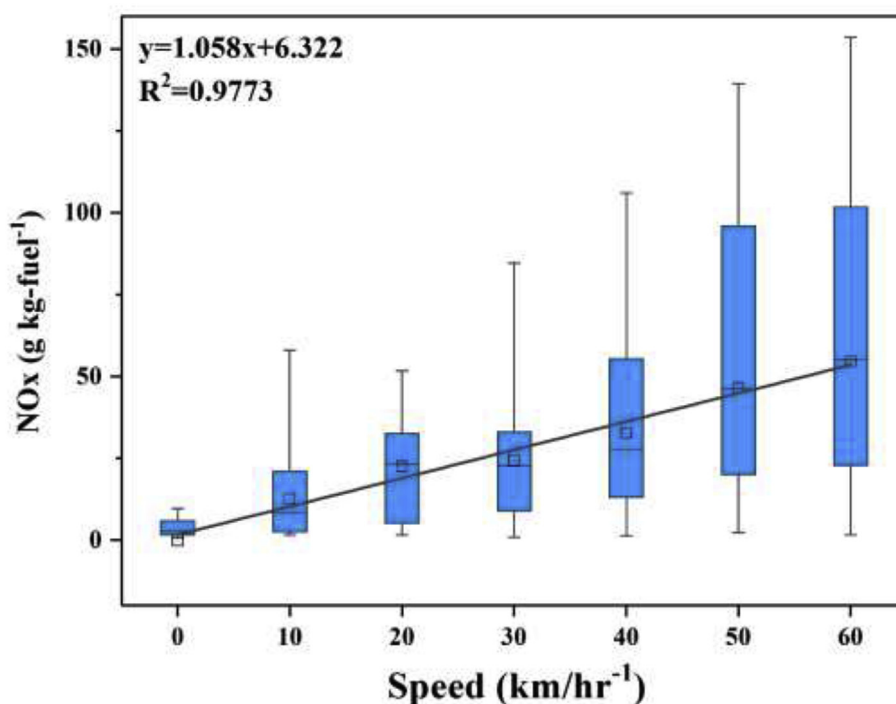
In the exhausts, alkanes were the most abundant VOCs ( $71.9 \pm 5.6\%$ ), followed by alkenes ( $25.2 \pm 5.4\%$ ), while ethyne ( $2.7 \pm 0.5\%$ ) and aromatic species ( $0.2 \pm 0.0\%$ ) were minor components. These percentages were consistent with those from other studies (Schifter et al., 2000; Chang et al., 2001; Lu et al., 2003; Wang et al., 2006; Adam et al., 2011), except that in Hong Kong where Guo et al. (2011) observed higher emission factors of aromatic species than alkynes.

As the main components of LPG fuel (Schifter et al., 2000), propane and butanes accounted for more than 90% of the alkane

**Table 1**  
Emission factors (kg kg-fuel<sup>-1</sup> for CO<sub>2</sub>, g kg-fuel<sup>-1</sup> for the others) of CO<sub>2</sub>, CO, NO<sub>x</sub>, VOC species and OFPs at different speeds for tested vehicles (n = 26).

Compounds	Driving speeds						
	idling	10 km h <sup>-1</sup>	20 km h <sup>-1</sup>	30 km h <sup>-1</sup>	40 km h <sup>-1</sup>	50 km h <sup>-1</sup>	60 km h <sup>-1</sup>
CO <sub>2</sub>	2.82 ± 0.291 <sup>a</sup>	2.87 ± 0.235	2.89 ± 0.202	2.87 ± 0.219	2.87 ± 0.226	2.84 ± 0.228	2.87 ± 0.211
CO	100 ± 176	76.5 ± 139	63.6 ± 117	74.2 ± 130	75.9 ± 131	91.8 ± 130	74.0 ± 121
NO <sub>x</sub>	4.13 ± 4.21	19.8 ± 28.5	31.9 ± 51.2	34.0 ± 45.9	44.4 ± 53.1	61.1 ± 58.5	71.1 ± 60.4
Alkyne							
Ethyne	0.201 ± 0.403	0.215 ± 0.422	0.193 ± 0.418	0.158 ± 0.368	0.142 ± 0.313	0.183 ± 0.383	0.172 ± 0.347
Sub-OFPs	0.187 ± 0.375	0.200 ± 0.393	0.179 ± 0.389	0.147 ± 0.342	0.132 ± 0.391	0.170 ± 0.356	0.160 ± 0.322
Alkenes							
Ethene	0.771 ± 1.30	1.11 ± 1.81	1.18 ± 1.93	1.23 ± 2.16	1.41 ± 2.87	0.953 ± 1.67	1.10 ± 1.75
Propene	0.297 ± 0.457	0.519 ± 0.758	0.515 ± 0.895	0.475 ± 0.893	0.627 ± 1.42	0.577 ± 0.923	0.524 ± 0.851
1-Butene ( × 10 <sup>3</sup> )	39.3 ± 69.4	101 ± 324	43.4 ± 80.3	41.6 ± 81.3	54.5 ± 133.9	42.5 ± 77.5	42.6 ± 74.5
Trans-2-butene ( × 10 <sup>3</sup> )	18.3 ± 46.6	84.2 ± 392	9.55 ± 21.4	8.63 ± 18.9	11.5 ± 28.7	8.33 ± 14.7	8.12 ± 14.9
Sub-total	1.13 ± 1.71	1.82 ± 2.55	1.75 ± 2.84	1.75 ± 3.06	2.10 ± 4.37	1.58 ± 2.49	1.68 ± 2.63
Sub-OFPs	10.8 ± 17.9	17.8 ± 33.4	16.8 ± 28.1	16.7 ± 30.2	20.1 ± 43.0	15.4 ± 26.1	16.1 ± 25.9
Alkanes							
Ethane	0.227 ± 0.319	0.189 ± 0.242	0.167 ± 0.205	0.168 ± 0.213	0.217 ± 0.267	0.201 ± 0.263	0.237 ± 0.385
Propane	4.09 ± 10.6	2.53 ± 3.89	2.46 ± 3.42	2.65 ± 4.32	3.16 ± 4.96	3.09 ± 4.79	2.62 ± 3.53
i-Butane	0.940 ± 2.43	0.558 ± 0.975	0.600 ± 0.853	0.598 ± 0.924	0.686 ± 1.04	0.693 ± 1.08	0.642 ± 0.912
n-Butane	1.63 ± 3.90	0.985 ± 1.73	1.02 ± 1.68	0.912 ± 1.47	1.01 ± 1.68	1.03 ± 1.77	0.977 ± 1.50
i-Pentane ( × 10 <sup>3</sup> )	21.3 ± 38.9	24.8 ± 56.9	23.3 ± 52.2	23.4 ± 57.5	27.7 ± 43.1	28.3 ± 55	25.6 ± 51.4
Sub-total	6.91 ± 17.2	4.29 ± 6.73	4.27 ± 6.08	4.35 ± 6.83	5.10 ± 7.87	5.04 ± 7.79	4.50 ± 6.11
Sub-OFPs	4.83 ± 12.1	2.96 ± 4.94	3.01 ± 4.51	2.98 ± 4.78	3.44 ± 5.45	3.43 ± 5.52	3.11 ± 4.48
Aromatics							
Benzene ( × 10 <sup>3</sup> )	5.70 ± 16.9	2.65 ± 9.91	3.60 ± 9.72	4.29 ± 15.0	5.92 ± 15.3	4.32 ± 12.9	3.29 ± 7.83
Toluene ( × 10 <sup>3</sup> )	4.44 ± 8.99	9.94 ± 35.5	10.6 ± 36.2	11.0 ± 39.9	7.74 ± 21.1	6.64 ± 25.1	9.17 ± 38.8
Sub-total ( × 10 <sup>3</sup> )	10.1 ± 23.0	12.6 ± 45.2	14.2 ± 44.7	15.3 ± 54.5	13.7 ± 35.8	11.0 ± 38.0	12.5 ± 45.6
Sub-OFP ( × 10 <sup>3</sup> )	21.2 ± 46.6	40.4 ± 144	43.7 ± 147	45.5 ± 165	34.1 ± 92.5	28.8 ± 106	37.8 ± 156
Total VOCs	8.24 ± 18.0	6.33 ± 9.24	6.23 ± 8.74	6.27 ± 9.36	7.36 ± 11.3	6.82 ± 10.2	6.36 ± 8.89
Total OFPs	15.8 ± 30.4	21.0 ± 38.8	20.0 ± 33.2	19.8 ± 35.4	23.8 ± 48.8	19.1 ± 32.1	19.4 ± 30.9

<sup>a</sup> Mean ± standard deviation.



**Fig. 1.** Changes in average emission factors of NO<sub>x</sub> (E<sub>NOx</sub>) with driving speeds. Box charts present the following order from bottom to the top: 10%, 25%, median (square in the box), average (line in the box), 75%, and 90%. The average emission factors of NO<sub>x</sub> at each speed were used to plot the linear regression line.

species and ~70% of total VOCs, as incompletely burned components emitted from the LPG taxis (Tang et al., 2007; Raslavicius et al., 2014). Propane was the dominant VOC species in the exhaust with an average emission factor ranging from 2.46 g kg-

fuel<sup>-1</sup> (20 km h<sup>-1</sup>) to 4.09 g kg-fuel<sup>-1</sup> (idling; Table 1), and it alone accounted for 39.5–49.6% of total VOC emissions. If converted to mileage-based emission factors (Table S3), it was approximately twice that reported in Hong Kong (Guo et al., 2011), and more than



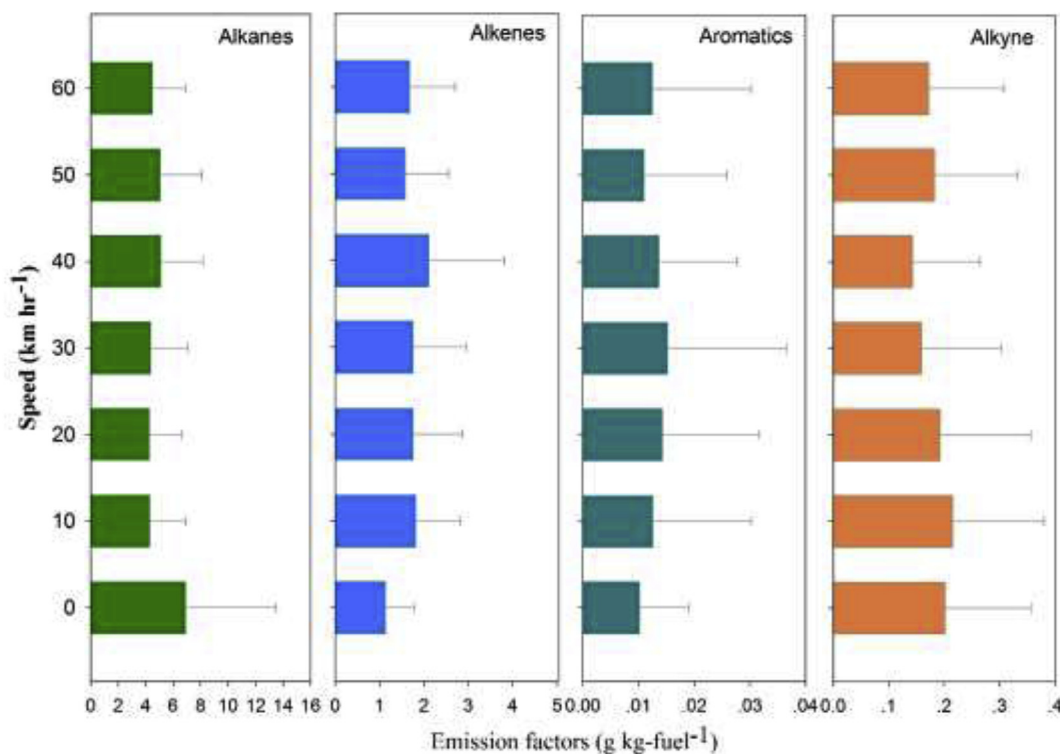


Fig. 2. Emission factors of VOC groups at idling and at constant speeds.

33 times of that reported in Varese, Italy (Adam et al., 2011), but ~28% lower than that reported in Taiwan (Chang et al., 2001).

Ethene accounted for more than 60% alkenes emitted from the LPG taxis. It was the second most abundant VOC species in the exhaust with average emission factors from 0.77 g kg-fuel<sup>-1</sup> (idling) to 1.41 g kg-fuel<sup>-1</sup> (40 km h<sup>-1</sup>; Table 1), accounting for 16.3% of the total emissions of VOCs. This fraction was consistent with that of 16.6% from a study in Varese, Italy (Adam et al., 2011) or 13.4% in Taiwan (Chang et al., 2001), while it was quite lower when compared to that of 0.6% in Hong Kong (Guo et al., 2011). Multi-step aliphatic hydrocarbon oxidation mechanisms proposed by Hautman et al. (1981) suggested that aliphatic hydrocarbons like propane or butanes can pyrolyze into ethene as the main intermediate product in high temperature before they are oxidized to CO or CO<sub>2</sub>. Therefore, without or with inefficient after-treatment devices, ethene might escape through the exhaust as an incomplete combustion product.

### 3.2.2. Influence of driving speeds on $E_{VOCs}$

Emission factors of the different VOC species from the tested LPG taxis at idling and constant speeds of 10–60 km h<sup>-1</sup> are shown in Table 1 and Fig. 3. It seemed that emission factors of alkanes and alkenes changed in a different pattern with the increase of speeds. Propane, as an example of alkanes, had a higher emission factor at idling (4.09 g kg-fuel<sup>-1</sup>), and its emission factors fluctuate from 2.53 g kg-fuel<sup>-1</sup> (20 km h<sup>-1</sup>) to 2.62 g kg-fuel<sup>-1</sup> (60 km h<sup>-1</sup>) when driving at constant speeds from 10 to 60 km h<sup>-1</sup>. On the contrary, alkenes had lower emission factors at idling than at cruising. Ethene, as an example, had lower emission factors at idling (0.77 g kg-fuel<sup>-1</sup>), and its emission factors increased with speed until it reached a peak at 40 km h<sup>-1</sup> (1.41 g kg-fuel<sup>-1</sup>), then dropped to 1.10 g kg-fuel<sup>-1</sup> at 60 km h<sup>-1</sup>. These findings were not consistent with the results from the study by Guo et al. (2011), who found that emission factors of either dominant VOC species, like propane and

ethene, or the total VOCs dropped dramatically from 25 km h<sup>-1</sup> to 50 km h<sup>-1</sup>, and then became stable at 70 and 100 km h<sup>-1</sup>. The different trends may be due to the difference in the combustion performance between gasoline-LPG bi-fuel LPG taxis in our study and dedicated LPG taxis in Hong Kong (Guo et al., 2011; Table S5), as a fuel-air mixture of gasoline-LPG bi-fuel engine is often too lean within the higher ranges of engine's rotational speed, which demonstrates the difficulties of ignition, flame extinction, and incomplete combustion and causes a huge increase in levels of emissions (Raslavicius et al., 2014). Moreover, Lyu et al. (2016) also pointed out that replacement of catalytic converter of LPG-fueled taxis in Hong Kong helped greatly in mitigating VOCs emission, so installation, replacement and maintenance of after-treatment devices are of great importance to reduce emissions.

### 3.3. Relationships between mileages and emission factors

The testes LPG-fueled taxis had model years from 1 to 5 years, and their mileages ranged (11.0–77.9) × 10<sup>4</sup> km. Average  $E_{NOx}$  showed an increasing trend with mileages ( $p < 0.05$ ) (Fig. 4) at a slope of ~0.56 g kg-fuel<sup>-1</sup> per 10<sup>4</sup> km of mileage. This increasing trend seemed to be more significant ( $p < 0.01$ ) between average  $E_{VOCs}$  and mileages. However, if we look further into the relationships between  $E_{VOCs}$  and mileages, average  $E_{VOCs}$  were found to be less than 2.0 g kg-fuel<sup>-1</sup> when mileages were below 40 × 10<sup>4</sup> km, while they varied from 0.258 to 30.2 g kg-fuel<sup>-1</sup> when mileages were above 40 × 10<sup>4</sup> km (Fig. 4). The variation in  $E_{VOCs}$  between these two mileage ranges may be affected by the maintenance of the vehicles (Chikhi et al., 2014; Tsai et al., 2017), which can be indicated by the CO/CO<sub>2</sub> ratio (Pierson et al., 1996; Chikhi et al., 2014). For LPG taxis with mileages below 40 × 10<sup>4</sup> km, the CO/CO<sub>2</sub> ratios were less than 5%, indicating good maintenance with lower VOC emissions; while for LPG taxis with mileages above 40 × 10<sup>4</sup> km, the CO/CO<sub>2</sub> ratios varied from 0.010–17.4%, indicating

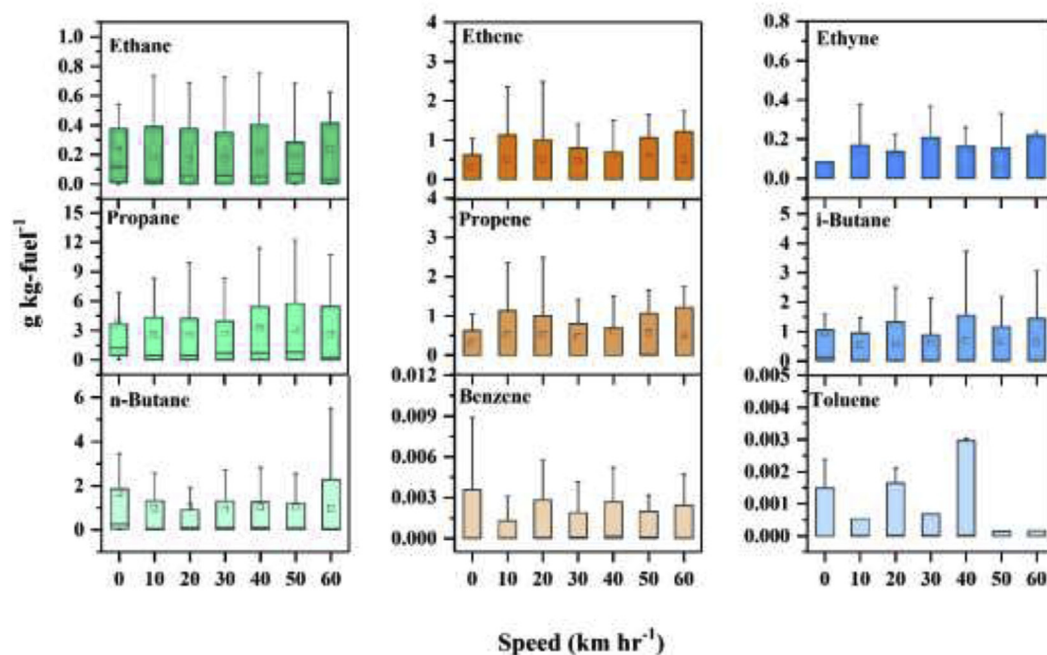


Fig. 3. Average emission factors of individual VOCs at idling and at constant speeds.

bad maintenance for some taxis with higher emissions of VOCs. Tsai et al. (2017) proposed that for gasoline motorcycles above a guaranteed mileage point, the engine and after-treatment system could deteriorate and cause increasing emissions. This study suggests that enhanced maintenance and inspection of LPG taxis at a certain guaranteed mileage, such as  $40 \times 10^4$  km in this study, would greatly reduce emissions of VOCs.

### 3.4. Environmental implications

LPG is widely recognized for its inherent environmental benefits, replacing a diesel/gasoline engine with an LPG-fueled equivalent would greatly reduce the emission of damaging particulates, reactive organic gases (ozone/SOA precursors) and toxic air pollutants (such as PAHs, benzene, and formaldehyde) in exhaust emission or during refueling. However, as indicated in this study, if engines are not LPG-dedicated and/or with poor exhaust after-treatment, these benefits would be discounted by higher  $\text{NO}_x$  emission and excessive propane/butanes in the exhaust.

From the tests in this study, the emission factors (EFs) of  $\text{NO}_x$  from LPG taxis increased linearly with speeds, therefore at a driving speed of  $35 \text{ km h}^{-1}$  the EF of  $\text{NO}_x$  could be estimated as  $2.69 \text{ g km}^{-1}$ , the average of that at  $30 \text{ km h}^{-1}$  ( $34.0 \text{ g kg}^{-1}$  or  $2.34 \text{ g km}^{-1}$ ; Table 1) and at  $40 \text{ km}^{-1}$  ( $44.4 \text{ g kg}^{-1}$  or  $3.05 \text{ g km}^{-1}$ ; Table 1). This average EF of  $\text{NO}_x$  ( $2.69 \text{ g km}^{-1}$ ) at  $35 \text{ km h}^{-1}$  is about twice of  $1.29 \text{ g km}^{-1}$  per vehicle for total car fleet at an average speed of  $\sim 35 \text{ km h}^{-1}$  based on tests in a busy urban tunnel in 2014 in Guangzhou (Zhang et al., 2015). It is worth noting that this study was mainly conducted in 2018, and EF of  $\text{NO}_x$  from gasoline or diesel cars should have decreased drastically in recent 5 years due to tightened emission limits with the updates of emission standards. A recent test in the same urban tunnel in November 2019 revealed that the average EF of  $\text{NO}_x$  for car fleet was  $0.24 \text{ g km}^{-1}$ , far below that of  $2.69 \text{ g km}^{-1}$  at a driving speed of  $35 \text{ km h}^{-1}$  as estimated from this study. In fact, this average EF is also much higher than the reported EFs of  $\text{NO}_x$  ranging  $0.010\text{--}1.1 \text{ g km}^{-1}$  (Table S2) for gasoline vehicles (Ntziachristos and Samaras, 2000;

Heeb et al., 2006; Pelkmans and Debal, 2006; André and Rapone, 2009; Adam et al., 2011; Hu et al., 2012; Yang et al., 2018); Although EFs of  $\text{NO}_x$  for LPG taxis ( $1.4\text{--}4.9 \text{ g km}^{-1}$  at  $10\text{--}60 \text{ km h}^{-1}$ ) are much lower than those of  $10.7\text{--}21.8 \text{ g km}^{-1}$  reported about 20 years ago for heavy trucks (Wang et al., 1997, 2000; Ramamurthy and Clark, 1999; Prucz et al., 2001), they are even much higher than those of  $0.52\text{--}1.7 \text{ g km}^{-1}$  reported for small or medium trucks when the LPG taxis are driving at  $40 \text{ km h}^{-1}$  or above (Joumard et al., 2003; André and Rapone, 2009; Chiang et al., 2012; Hu et al., 2012; Chikhi et al., 2014). Therefore, the transition of a gasoline taxi to a taxi with an LPG-gasoline bi-fuel engine and without or with inefficient exhaust after-treatment would not benefit the emission control of  $\text{NO}_x$ .

LPG is mainly composed of propane and butanes, which are far less reactive and toxic when compared to the constituents of gasoline, therefore replace gasoline-powered taxis with LPG taxis are supposed to lower the ozone formation potentials (OFPs) and hazardous air pollutants in the exhaust. From the study air toxics like aromatic hydrocarbons did become trivial in the exhaust of LPG taxis. As for the OFPs calculated based on the maximum increment reactivity (MIR) method developed by Carter (1994, 2009), they ranged  $19.1\text{--}23.8 \text{ g kg-fuel}^{-1}$ , or  $1.3\text{--}1.6 \text{ g km}^{-1}$ , when cruising at speeds from 10 to  $60 \text{ km h}^{-1}$ , much higher than that at idling ( $15.8 \text{ g kg-fuel}^{-1}$  or  $1.1 \text{ g km}^{-1}$ ) largely due to higher emission of more reactive alkenes when cruising (Table 1). This is quite different from the results for gasoline vehicles, which have much more emissions of reactive alkenes and aromatics at idling than at cruising (e.g. Zhang et al., 2020). Similarly, if we estimate the EF and the OFP at  $35 \text{ km h}^{-1}$  as the average of that at  $30 \text{ km h}^{-1}$  and  $40 \text{ km h}^{-1}$  (Table 1), LPG taxis at  $35 \text{ km h}^{-1}$  would have an average EF of VOCs as  $6.82 \text{ g kg-fuel}^{-1}$  or  $0.47 \text{ g km}^{-1}$ , similar to that of  $0.45 \text{ g km}^{-1}$  per vehicle tested in the Zhujiang Tunnel for vehicle fleet in urban Guangzhou (Zhang et al., 2018a). However, the estimated OFPs of VOCs from LPG taxis at  $35 \text{ km h}^{-1}$ ,  $21.8 \text{ g kg-fuel}^{-1}$  or  $1.5 \text{ g km}^{-1}$ , was 1.5 times of that  $1.0 \text{ g km}^{-1}$  reported for vehicle fleet based on the tunnel tests in 2014 (Zhang et al., 2018a). It is also worth noting that based on tests in the same Zhujiang Tunnel, average OFPs of

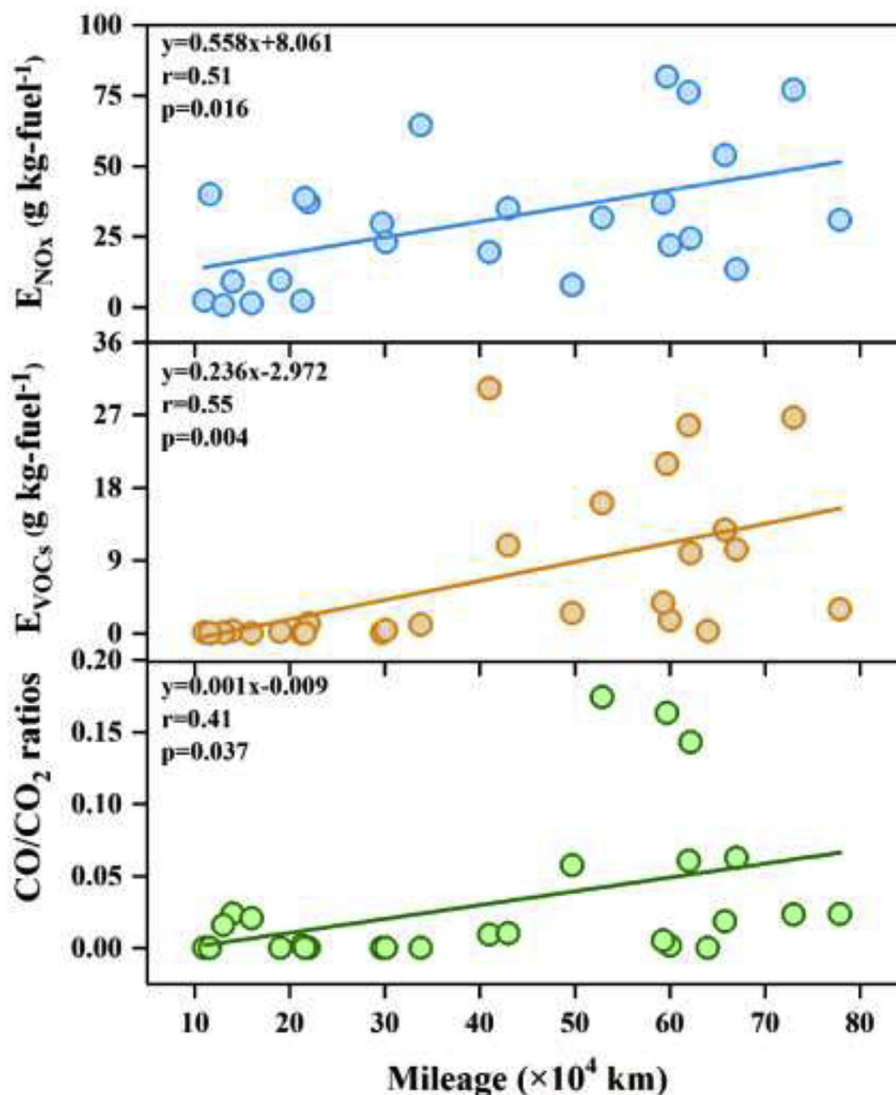


Fig. 4. Correlations of between mileage and NO $_x$  emission factor, VOCs emission factor and CO/CO $_2$  ratio of LPG-fueled taxis.

VOCs in terms of mg km $^{-1}$  per vehicle decreased by ~50% during 2004–2014, the decrease rate from 2014 to 2018 should be more rapid due to update of fuel quality standards and vehicle emission standards. Therefore, replacing gasoline taxis with the LPG-gasoline bi-fuel taxis without or with inefficient exhaust after-treatment did not show the potentials to reduce reactive VOCs as ozone precursors, largely due to relatively higher emission of reactive alkenes (Table 1). In fact, we can also compare the emissions of reactive VOCs in terms of OFPs with those for on-road vehicles based on tunnel tests in other parts of the world (Table S4). Tested OFPs for the LPG taxis in this study ranged 1.1–1.6 g km $^{-1}$ ; they were also much higher when compared to that of 0.15 g km $^{-1}$  per vehicle as observed in the Leopold II Tunnel in Brussels, Belgium for vehicle fleet without LPG cars (Ait-Helal et al., 2015), 0.18 g km $^{-1}$  per vehicle as observed in the Shing Mun Tunnel in Hong Kong (Cui et al., 2018), 0.74 g km $^{-1}$  per vehicle observed in the Hsuehshan tunnel in Taiwan (Liu et al., 2014c), or even 0.44 g km $^{-1}$  per vehicle observed in the Fu Gui Mountain tunnel in Nanjing (Zhang et al., 2018b), and only ~20% lower than that of 1.93 g km $^{-1}$  per vehicle observed in Loma Larga Tunnel in Monterrey, Mexico (Araizaga et al., 2013).

The unsatisfactory performance of the LPG taxis in the

emissions of ozone precursors (NO $_x$  and reactive VOCs) is rooted in the fact that the LPG taxis in this study are LPG-gasoline bi-fuel ones without catalytic converters or with inefficient ones. If the LPG-gasoline bi-fuel taxis are ones with dedicated-LPG engines and efficient after-treatment devices as did in Hong Kong, their emission of NO $_x$  and VOCs would be largely reduced (Guo et al., 2011; Lau et al., 2011; Lyu et al., 2016).

#### 4. Conclusions

In this study, 26 LPG-fueled taxis in Guangzhou were tested at idling or cruising at speeds from 10 to 60 km h $^{-1}$  using a chassis dynamometer for their emissions of NO $_x$  and VOCs, both of which are ozone precursors. The results revealed that the  $E_{NO_x}$  increased with speed steadily from  $4.13 \pm 4.21$  g kg-fuel $^{-1}$  to  $71.1 \pm 60.4$  g kg-fuel $^{-1}$  by approximately 10.6 g kg-fuel $^{-1}$  on average for every 10 km h $^{-1}$  increase in speed;  $E_{VOC_s}$  of the LPG taxis were higher at idling (8.24 g kg-fuel $^{-1}$ ) than when cruising at 10–60 km h $^{-1}$  (6.22–7.36 g kg-fuel $^{-1}$ ), but the OFPs at idling (15.8 g kg-fuel $^{-1}$ ) was lower than that of 19.1–23.8 g kg-fuel $^{-1}$  when cruising at speeds from 10 to 60 km h $^{-1}$ , largely due to higher emission of more reactive alkenes when cruising. These results also suggested that



replacing the gasoline-power taxis with the LPG-powered ones with LPG-gasoline bi-fuel engines and inefficient after-treatment is not a successful approach to reduce the emission of NO<sub>x</sub> and reactive VOCs, and therefore would not benefit the surface ozone pollution control, and instead replacing the gasoline-powered taxis with the LPG-ones with dedicated LPG engines and efficient after-treatment converters, as the case in Hong Kong, would greatly lower the emissions of reactive VOCs and NO<sub>x</sub>. Our study also revealed that when the *E*<sub>VOCs</sub> were found to be less than 2.0 g kg-fuel<sup>-1</sup> when mileages were below 40 × 10<sup>4</sup> km, otherwise they ranged 0.258–30.2 g kg-fuel<sup>-1</sup>, suggesting that enhancing the maintenance and inspection might help curb emissions of VOCs when mileages were above a guaranteed mileage point.

It is worth noting that we only investigated emissions under idle and cruising modes. We did not test emissions during criteria driving cycles or conducted on-board emission measurements for real-world driving, and we did not further test emissions under the acceleration and deceleration modes, largely due to that we wanted to measure specific VOC species to assess their OFPs. However, as indicated in previous studies (e.g., Lau et al., 2011), emissions of hydrocarbons from LPG taxis under acceleration and deceleration modes might be near that under cruising mode, and the emissions of NO<sub>x</sub> under cruising might lay between that under the acceleration and deceleration modes, therefore the conclusion that the studied LPG taxis do not benefit ambient ozone pollution control still stands.

#### Authorship Contribution Statement

Jingjing Feng: Writing - original draft, Methodology, Formal analysis, Writing - review & editing. Yanli Zhang: Conceptualization, Writing, Methodology, Formal analysis, Writing - review & editing. Wei Song: Investigation, Data curation, Resources. Wei Deng: Investigation, Data curation, Resources. Ming Zhu: Investigation, Data curation, Resources. Zheng Fang: Investigation, Data curation. Yuqing Ye: Investigation, Data curation. Hua Fang: Investigation, Data curation. Zhenfeng Wu: Investigation, Data curation. Scott Lowther: Writing - review & editing. Kelvin C. Jones: Writing - review & editing. Xinming Wang: Project administration, Conceptualization, Writing, Funding acquisition, Supervision- review & editing.

#### Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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#### Appendix A. Supplementary data

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