

# Experimental constraints on the origin of potassium-rich adakites in eastern China

## 中国东部富钾埃达克岩成因的实验约束\*

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**Abstract** Adakites are geochemically distinct volcanic and plutonic granitoid rocks found in intraoceanic island arc settings where relatively young, hot oceanic lithosphere has been subducted and melted (termed "slab melting"), and in continental arcs, such as the Andes, where melting has taken place at the base of tectonically- or magmatically-thickened lower crust (termed "lower crustal melting"). In both settings, the distinctive geochemical signature of adakitic granitoids is attributed to an origin by partial melting of a variably-hydrated metabasaltic protolith at sufficient depths for garnet to be stable within the residual crystalline assemblage (i.e., residues of garnet-amphibolite and/or eclogite). Once generated, "pristine" or "parental" adakite melts may have their composition subsequently modified by processes of assimilation (of either mantle or continental material) and crystal fractionation during transport to and emplacement in the middle-upper crust. Late Mesozoic (early-mid Cretaceous, ~ 160-110 Ma) adakites in eastern China are unusually rich in potassium (K<sub>2</sub>O) and other large-ion lithophile elements (e.g., Ba, Th, U), with low Na<sub>2</sub>O/K<sub>2</sub>O ratios (~ 1.0-1.1), in contrast to sodic adakites, found in eastern China and elsewhere, which resemble experimental adakite liquids produced by dehydration melting of basalt in the garnet-amphibolite to eclogite facies and which formed by either slab melting of oceanic crust, or by partial melting of broadly basaltic, lower crustal protoliths. Despite these compositional differences, their overall geochemical character defines the potassic granitoids of eastern China as adakites. We attribute the unique chemistry of these potassium-rich adakites to either peculiarities in the composition of their source, or to the processes, including assimilation and fractional crystallization (AFC), that subsequently modified parental adakite magmas. Although the apparent lack of proximity to a subduction zone suggests that adakites in eastern China formed by partial melting of underplated (magmatically-thickened) mafic lower crust, geodynamic scenarios involving "flat slab" subduction in eastern China during the Yanshanian period cannot be ruled out.

**Key words** Potassium-rich adakites, Experimental constraints, Eastern China

**摘要** Adakite 在地球化学上具明显特征的火山岩和深成花岗岩类岩石, 见于洋内岛弧环境和大陆弧, 如安底斯弧。在洋内岛弧, 由热的消减的年轻大洋岩石圈熔融形成(叫做"板片熔融"), 而在大陆弧, 熔融曾发生在构造或岩浆加厚的下地壳底(叫做"下地壳熔融")。在这两种产状环境中, adakite 的鲜明地球化学特征被认为是起因于, 一种不同程度含水的变质基性原岩在足够深度上的部分熔融; 这里的足够深度是指可使石榴子石在残余结晶组合(即石榴角闪石和/或榴辉石的残余)中保持稳定的深度。"原始"或"母" adakite 熔体一旦形成, 便可能在其向上迁移和侵位到中上地壳期间受到同化作用(或是地幔, 或是大陆物质)和结晶分异作用的改造。中国东部晚中生代(早中白垩世, 160-110 Ma)的 adakite, 与见于同一地区和其它地方的钠质 adakite 相比, 通常富含钾(K<sub>2</sub>O)和其它大离子亲石元素(如 Ba, Th, U), 有较低的 Na<sub>2</sub>O/K<sub>2</sub>O 比值(~ 1.0-1.1), 类似于

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玄武岩在石榴角闪岩-榴辉岩相含水熔融实验中所产生的 adakite 熔体, 要么是由洋壳板片熔融所形成, 要么是由不同成分的玄武质下地壳原岩部分熔融所形成。尽管有这些成分差异, 它们的总体化学特征仍然可将中国东部的富钾花岗岩类岩石定为 adakite。我们把这些富钾的 adakite 的独特化学特征, 归因于成分来源的特殊性, 或 adakite 母岩浆遭受了同化混染和结晶分异 (AFC) 作用的改造。虽然中国东部与消减带环境明显不同这一点表明, 那里的 adakite 可由板块底部侵位的 (岩浆加厚的) 镁铁质下地壳部分熔融所形成, 但燕山运动期间中国东部存在“平坦”俯冲的地球动力学环境是可能被排除的。

关键词 富钾 adakite; 实验制约; 中国东部

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## Introduction

### Adakites as slab melts

For the past decade, the term adakite has been used to describe arc-related volcanic and plutonic rocks of andesitic-dacitic composition, possessing distinctive geochemical characteristics, including high  $\text{Al}_2\text{O}_3$  and  $\text{Na}_2\text{O}$  contents, and high  $\text{La}/\text{Yb}$  and  $\text{Sr}/\text{Y}$  ratios coupled with strong depletions in  $\text{Yb}$ ,  $\text{Y}$ , and high-field strength elements (“HFSEs”, including  $\text{Ti}$ ,  $\text{Nb}$ ,  $\text{Ta}$ ), suggestive of an origin by partial melting of basaltic ocean crust within downgoing slabs of oceanic lithosphere in subduction zones (Defant and Drummond, 1990; Drummond and Defant, 1991). Many of these key geochemical features were attributed to the presence of garnet in the residues of melting, and adakite petrogenesis was viewed as resulting from partial melting accompanying the dehydration of subducted oceanic crust, variably metamorphosed to a garnet-amphibolite and/or eclogite facies assemblage. Thus adakites occurring along convergent plate boundaries (in both intraoceanic and continental arc settings) were interpreted to be the products of “slab melting” (Defant and Drummond, 1990). Numerical models for the thermal regime in subducting slabs (Peacock *et al.*, 1996; Molnar and England, 1995), however, suggested that slab melting would generally be restricted to unusually “hot” subduction zones, for example, those associated with subduction of an oceanic ridge (*e.g.*, Stern and Killian, 1996; Kay *et al.*, 1993), or of very young oceanic lithosphere (Sajona *et al.*, 1994; Kepezhinskis *et al.*, 1995; Morris, 1995); slow, oblique subduction, as in the western Aleutian arc (Kay, 1978; Yagodzinski *et al.*, 1995), or subduction along a “slab tear”, as in the Kamchatkan arc (Yagodzinski *et al.*, 2001).

The geochemical basis for the “slab melting” model for adakite petrogenesis came from experimental studies that showed that 10% ~ 30% melting of variably-hydrated MORB-like basalts at pressures of 1 - 4 GPa produced sodium-rich granitoid liquids that were, in terms of their major element components, similar to adakites from a number of continental and island arc settings (*e.g.*, high  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$  and  $\text{Na}_2\text{O}$  contents;  $\text{Na}_2\text{O}/\text{K}_2\text{O}$  ratios  $> 2.0$ ; low  $\text{MgO}$  and  $\text{K}_2\text{O}$  contents, and  $Mg^\# < 0.5$ , where  $Mg^\# = \text{molar} [\text{Mg}/(\text{Mg} + \text{Fe})]$ ) (Rapp *et al.*, 1991; Winter and Newton, 1991; Sen and Dunn, 1994; Rapp and Watson,

1995; Rapp, 1995;). Trace element geochemical features common to adakites worldwide, including strong enrichments in large-ion lithophile elements (LILEs), high ratios of  $\text{La}/\text{Yb}$  and  $\text{Sr}/\text{Y}$  coupled with strong relative depletions in heavy rare-earth elements (HREE),  $\text{Y}$ , and high field-strength elements (HFSEs, *e.g.*,  $\text{Ti}$ ,  $\text{Nb}$  and  $\text{Ta}$ ) were, as a consequence, explicable in terms of equilibration between adakite liquids and garnet-bearing crystalline residues (eclogite or garnet amphibolite, with or without minor phases such as quartz or plagioclase); the presence of accessory titanate phases, such as sphene or rutile, are required to impart the strong relative depletion in HFSEs observed in adakites.

Although compositionally “pristine” adakites were described from some subduction zones (*e.g.*, SW Japan, Morris, 1995; western Panama, Defant *et al.*, 1991), in other cases, adakite magmas formed by slab melting appear to have undergone variable extents of interaction with rocks within the mantle wedge and lower arc crust (*e.g.*, in the southern Andes, Stern and Killian, 1996 and Kay *et al.*, 1993; in the Aleutians, the type-locality, Kay, 1978; and in the Philippines, Sajona *et al.*, 1994), with the effects of these interactions manifested in the geochemical and isotopic attributes of “hybridized” adakites. Trace element analyses of “pristine” and “mantle-hybridized” adakitic liquids produced in laboratory experiments at 3~4 GPa (Rapp *et al.*, 1999; this study) confirm the general legitimacy of petrologic models for adakite petrogenesis that begin with partial melting of a hydrous, mafic (*i.e.*, basaltic) protolith in the garnet-amphibolite to eclogite facies. Limited interaction with peridotite in the overlying mantle wedge produces Mg-rich, hybridized liquids which retain the characteristic “adakitic” geochemical signature.

### Adakites as crustal melts

An alternative to the slab melting model for adakite petrogenesis was proposed by Atherton and Petford (1993), to explain the origin of Na-rich plutonic and volcanic granitoids (*i.e.*, ignimbrites, quartz diorites and tonalites) from the Cordillera Blanca complex in the Andes of NW Peru. At the time of adakite magmatism, the subducting slab at the trench (the Nazca Plate) was 55~65 Myr old, and therefore presumably “too old and too cold” to undergo melting; instead, it was suggested that the Cordillera Blanca granitoids were the products of partial melting of mafic lower crust at the base of the magmatically-thickened keel.

of the Andes (Atherton and Petford, 1993; Petford and Atherton, 1996), similar to the model put forth by Smith and Leeman, (1987) for the petrogenesis of Na-rich dacites at Mt St Helens in the Cascade Range of North America. Adakite-like rocks from South Island, New Zealand (Muir *et al*, 1995), the Antarctic Peninsula (Wareham *et al*, 1997), and the Klamath Mountains of California and Oregon (Barnes *et al*, 1992; 1996) were subsequently interpreted as the products of partial melting of mafic lower crust, formed in response to crustal thickening achieved through underplating of mantle-derived mafic magmas (emplaced at the base of the existing arc crust). It was argued that displacement of this over-thickened "keel" of mafic crust to depths sufficient for garnet stability had led to partial melting attendant to dehydration, and adakite magmatism. Recent numerical models suggest that the appropriate *P-T* conditions can be achieved for partial melting of underplated "mafic lower (arc) crust" to take place, generating adakite parent magmas with mixed mantle- and crustal-like geochemical and isotopic signatures (Petford and Gallagher, 2001).

Intermediate-felsic volcanic and plutonic rocks of late-Cretaceous age (160- 110 Myr old) from eastern China possess all of the characteristic geochemical attributes of adakites, including high  $SiO_2$  and  $Al_2O_3$  contents and high La/Yb and Sr/Y ratios, and strong relative depletions in the HREEs, Y, and HFSE. Yet many of the adakites in eastern China are distinctly more potassic than typical sodium-rich adakites associated with subduction, with Na<sub>2</sub>O/K<sub>2</sub>O ratios close to 1 (see Pan *et al*, 2001; Wang and Zhang, 2001; Wang *et al*, 2001; Zhang *et al*, 2001), and they are generally more potassic than the more typically sodic adakites from the Cascades, Antarctica, and the Andes. The "adakite-like" or "adakitic" granitoids which occur throughout eastern China have been termed continental or "C-type" adakites (Zhang *et al*, 2001), because they appear to lack any temporal or spatial association with subduction, and a lower crustal melting origin has been proposed (Wang and Zhang, 2001; Pan *et al*, 2001; Wang *et al*, 2001). "C-type" adakites are contrasted with more sodic "oceanic" or "O-type" adakites (with Na<sub>2</sub>O/K<sub>2</sub>O ratios > 2.0) possessing a clear association with subduction (e.g., Xu *et al*, 2000; Xu *et al*, 2001). If the potassium-rich adakites of eastern China are also derived by partial melting of a garnet-bearing, hydrous mafic source, then their compositional differences with "O-type" adakites must be due either to differences in the source, or in the processes (e.g., assimilation and fractional crystallization; AFC) that subsequently modified the composition of "pristine" or parental adakite magmas. Melting experiments on hydrous basalt provide a geochemical reference point from which the geochemical effects of melt-rock reaction, wall rock assimilation, and intracrustal fractionation on adakite composition can be assessed.

## Review of Experimental Studies

Whether formed by slab melting or lower crustal

melting, a number of factors will control the composition of "parental" adakite magmas. These include (1) the bulk composition of the source basalt, (2) the pressure-temperature conditions of melting, and (3) the exact mineralogy of the crystalline residue, especially the presence or absence of minor and accessory phases (e.g., rutile, zircon, or apatite) that can strongly affect the distribution of certain trace elements.

### Basalt compositions used in melting experiments

A variety of both natural and synthetic starting materials have been used to study the high-pressure melting behavior of hydrous basalt (Table 1). Rapp *et al* (1991) and Rapp and Watson (1995) conducted melting experiments at 0.8- 3.2 GPa and 1000 - 1150 °C on four different natural amphibolites compositionally representative of variably altered and metamorphosed MORB. Potassium content in these starting materials ranges from < 0.1 wt% to 0.8 wt% K<sub>2</sub>O, and Al<sub>2</sub>O<sub>3</sub> contents range from 14.2 wt% to 17.0 wt% (Table 1). Adakitic liquids were produced by dehydration melting reactions involving hydrous minerals (primarily amphibole) whose breakdown produces silica-rich liquids at relatively low- to moderate-degrees of melting. Dehydration melting experiments were also conducted by Sen and Dunn (1994) at 1.5 and 2.0 GPa, temperatures between 800 and 1150 °C, using a natural high-K amphibolitic basalt as starting material (Table 1), and Winter and Newton (1991) conducted melting experiments over a range of pressures (0.5- 3.0 GPa) and temperatures (850 - 1000 °C), using two different low-K tholeiites (one a natural high-Al basalt, and the other a synthetic low-Al basalt representing "average Archean tholeiite") and variable amounts of water (from < 1.0% to 15% H<sub>2</sub>O).

Despite the fact that the various starting materials used in these experimental studies possessed a broad range in K<sub>2</sub>O (0.1- 0.8 wt%), Na<sub>2</sub>O (2.2- 4.3 wt%), and Al<sub>2</sub>O<sub>3</sub> (14.2 - 19.3 wt%) contents, and highly variable Na<sub>2</sub>O/K<sub>2</sub>O (3 - 28), they all produced Na-rich granitoid liquids closely comparable to natural adakite by low- to moderate-degrees of melting (~ 10- 30% by weight), leaving generally eclogitic crystalline residues. Note however, that the Na<sub>2</sub>O/K<sub>2</sub>O ratio of average basalt from China is much lower than any of the starting materials used in the basalt melting experiments, with a K<sub>2</sub>O content (2.51 wt%) much higher than the worldwide average (1.10 wt%).

### Melting phase relations of hydrous basalt to 5 GPa

The melting phase diagram for hydrous basalt (Fig. 1) represent a composite constructed from the results of a number of experimental studies, on both natural and synthetic starting materials, and with variable amounts of water present. Several of these studies focused on sub-solidus phase relations and the stability limits of important hydrous minerals, such as amphibole, lawsonite, zoisite and clinozoisite, and phengite (Poli and Schmidt, 1995; Schmidt, 1996; Kerrick and Connolly, 2001; Lopez and Castro, 2001). The actual position of the solidus for a given bulk composition will be determined by the form and

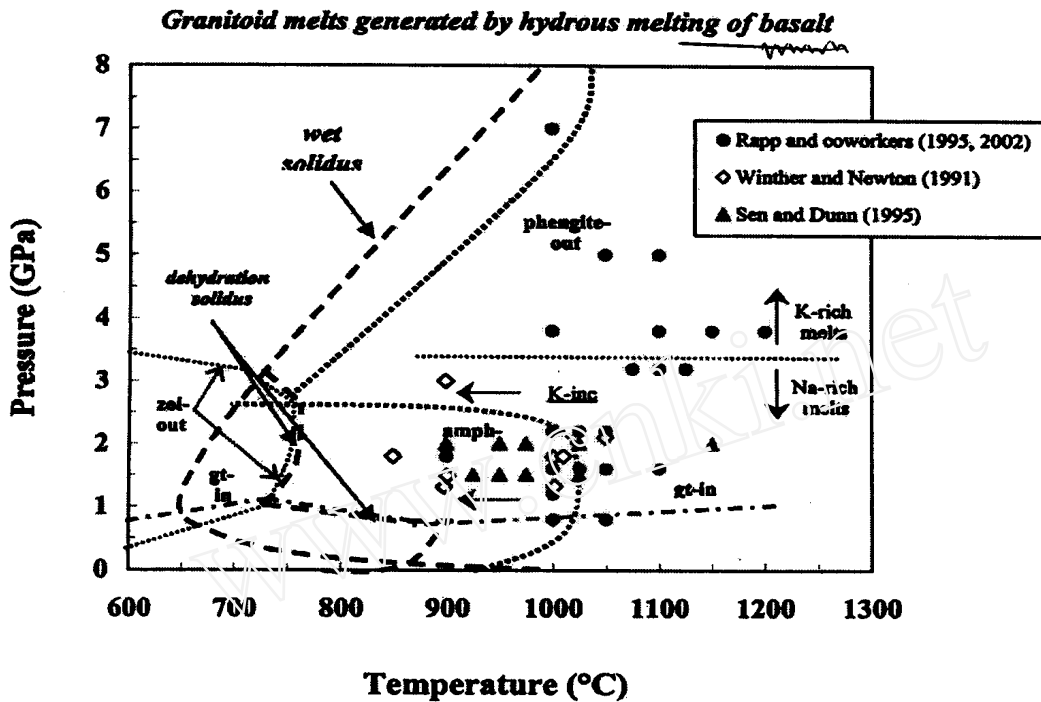


Fig. 1 Composite phase diagram for basalt-H<sub>2</sub>O, with *P-T* locations of melting experiments in which adakitic granitoid liquids were formed (see references in text), the water-saturated or "wet basalt" solidus and the water-absent or dehydration melting solidus, and major phase boundaries for garnet and hydrous minerals. At pressures less than 3.5 GPa, liquids formed by very low degrees of melting (< 10%) near the solidus have low Na<sub>2</sub>O/K<sub>2</sub>O ratios (< 1.5); adakitic liquids with higher Na<sub>2</sub>O/K<sub>2</sub>O ratios form at higher temperatures and greater degrees of melting (10-30%). At higher pressures (> 3.5 GPa), granitoid liquids formed by low- to moderate-degrees of melting possess high SiO<sub>2</sub> and K<sub>2</sub>O contents, very low Na<sub>2</sub>O/K<sub>2</sub>O ratios, and low Al<sub>2</sub>O<sub>3</sub> contents (see Table 2)

Table 1 Major element oxide composition of starting material in melting experiments on hydrous basalts (wt%)

| basalt                         | 1     | 2     | 3     | 4     | 5     | 6     | 7     | 8     |
|--------------------------------|-------|-------|-------|-------|-------|-------|-------|-------|
| SiO <sub>2</sub>               | 51.19 | 48.60 | 48.30 | 47.60 | 48.20 | 49.20 | 47.09 | 48.28 |
| TiO <sub>2</sub>               | 1.18  | 2.06  | 0.72  | 1.19  | 0.85  | 1.19  | 1.23  | 2.21  |
| Al <sub>2</sub> O <sub>3</sub> | 16.62 | 17.03 | 15.30 | 14.18 | 19.30 | 15.60 | 15.07 | 14.99 |
| FeO*                           | 11.32 | 10.69 | 10.70 | 13.77 | 8.94  | 13.60 | 13.20 | 11.13 |
| MnO                            | 0.23  | 0.21  | 0.19  | 0.19  | 0.11  | 0.15  | 0.26  | 0.20  |
| MgO                            | 6.59  | 6.07  | 8.40  | 6.86  | 8.72  | 6.89  | 8.29  | 7.00  |
| CaO                            | 5.49  | 9.66  | 12.60 | 10.99 | 10.50 | 10.80 | 11.33 | 8.07  |
| Na <sub>2</sub> O              | 4.33  | 3.30  | 2.27  | 2.56  | 3.11  | 2.35  | 2.52  | 3.40  |
| K <sub>2</sub> O               | 0.82  | 0.21  | 0.08  | 0.19  | 0.39  | 0.26  | 0.80  | 2.51  |

availability of water in the source; if a free hydrous fluid phase is present, then the source can be considered to be "fully hydrated", and melting will commence at the "wet basalt" solidus. If no free fluid phase is present, all water will be bound in hydrous minerals, and melting will commence at the "fluid-absent" solidus through dehydration melting reactions involving amphibole, zoisite, and other water-bearing phases. The solidus for dehydration melting occurs at somewhat higher temperatures than the water-saturated or wet basalt solidus, with its exact location being

determined by the relative degree of hydration of the source, and the proportion of hydrous phases present (Lopez and Castro, 2001; Vielzeuf and Schmidt, 2001).

#### Melt compositions at 1- 4 GPa

Also shown in Fig. 1 are the *P-T* locations of experiments in which granitoid melts (that is, liquid compositions that contain more than 10% quartz in a CIPW normative calculation) were reported (Winther and Newton,

1991; Sen and Dunn, 1994; Rapp and Watson, 1995). These experiments were generally conducted under fluid-absent conditions, where melting proceeds via dehydration reactions involving hydrous phases, primarily amphibole. Because potassium and the LLEs behave generally as highly incompatible elements during partial melting, the most potassic liquids, those richest in LLEs, will form at the lowest degrees of melting, nearest the solidus, and liquids formed by progressively higher degrees of melting will be less potassic and increasingly dilute with respect to the LLEs. Partial melts formed by very low degrees of melting (< 10%), will thus be more potassic, with lower Na<sub>2</sub>O/K<sub>2</sub>O ratios (0.9- 1.2), than their sodium-rich counterparts (Na<sub>2</sub>O/K<sub>2</sub>O > 1.5- 2.0), formed at slightly higher temperatures and higher degrees of melting (10- 30 wt%). The amount of K<sub>2</sub>O in the melt at a given degree of melting also appears to increase with increasing pressure (W inther and New ton, 1991), so that 20% melting at 3.8 GPa produces a more potassic liquid than 20% melting of the same basalt at 1.8 GPa (see Table 1). The lowest degrees of melting should also produce liquids with the highest Sr/Y and La/Yb ratios.

The amphibole-out phase boundary exerts strong control over the alumina saturation index (ASI, or A/CNK = molar Al<sub>2</sub>O<sub>3</sub>/(CaO + Na<sub>2</sub>O + K<sub>2</sub>O)) of the partial melt, such that metaluminous adakitic liquids (*i.e.*, A/CNK < 1.0- 1.1) are only produced close to or beyond this boundary (Rapp, 1995). Most adakites, whether in a subduction zone or lower crustal setting, are metaluminous, and therefore they probably formed at temperatures at which residual amphibole was either a minor phase, or not present at all, and an eclogitic phase assemblage (garnet and clinopyroxene) was predominant. In this regard, we note that most of the adakites from eastern China possess A/CNK ratios less than 1.1 (Zhang *et al.*, 2001).

## Geochemical Comparisons with C-type Adakites

Geochemical comparisons between the experimental adakitic liquids, adakites from eastern China, and adakites formed either by slab melting or lower crustal melting of subducted slabs reveal important similarities and differences, as outlined below, providing some initial constraints on the petrogenesis of the compositionally-distinctive potassic adakites.

### Major-element comparisons

Like "pristine" adakitic liquids in the basalt melting experiments, potassic adakites are typified by high SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> contents, Mg<sup>#</sup>'s less than 40- 45 (where Mg<sup>#</sup> = 100 x molar (Mg/[Mg + Fe])), and A/CNK ratios generally < 1.0- 1.1. Adakites related to slab melting often have higher Mg<sup>#</sup>'s at lower SiO<sub>2</sub> contents, and A/CNK ratios less than 1.0, reflecting varying extents of interaction with peridotite in the overlying mantle wedge, but trace element abundance patterns remain largely unaffected (see discussion in Rapp *et al.*, 1999, and Smithies, 2001). Clear

distinctions can therefore be made between relatively "pristine" and "mantle-hybridized" adakitic magmas, and although it is premature to call potassic adakites "pristine", it seems clear that they have experienced little interaction with a mantle component (as attested to by their low Mg<sup>#</sup>'s and low Cr and Ni contents). As shown by Wang *et al.* (2001) and Zhang and Wang (2001), the primary difference, in terms of major elements, between potassic adakite (C-type) and sodic adakite (O-type) is that the former is characterized by higher K<sub>2</sub>O contents and lower Na<sub>2</sub>O/K<sub>2</sub>O ratios over a comparable range of SiO<sub>2</sub> contents (60- 73 wt%). When plotted on a ternary feldspar (Ab-An-Or) diagram, potassic adakites fall close to or within the granodiorite and granite fields, relative to the more typically tonalitic-trondhjemitic compositions of sodic adakites, as well as most experimental adakite liquids (Rapp *et al.*, 1991; W inther and New ton, 1991; Sen and Dunn, 1994), with the exception of those produced from very low degrees of melting (*i.e.*, < 10 wt%).

### Trace element comparisons

Trace-element data for experimental liquids is relatively scarce; ion microprobe data has been published for adakitic liquids produced at 1- 3 GPa (analyzing several samples from Rapp and Watson, 1995), and both "pristine" and "mantle-hybridized" adakitic liquids at 3- 4 GPa (Rapp *et al.*, 1999). This data is combined with additional unpublished data (Rapp and Shimizu, in prep.) in making trace element comparisons between experimental adakitic liquids, potassium-rich adakites from China, and Na-granitoids of the Cordillera Blanca batholith (to typify adakites derived by partial melting of thickened crust). Plots of La/Yb versus Yb and Sr/Y and Y were first used by Martin (1987) and Defant and Drummond (1990) to discriminate adakites and their Archean analogues (TTG) from normal calc-alkaline andesite-dacite-rhyolite volcanics. Mantle-normalized trace element abundance patterns for liquids formed at 3- 4 GPa, from melting of two different basalts, and taken to be representative of pristine adakite (unpublished data of Rapp and Shimizu), are used to make broader geochemical comparisons.

#### Sr/Y versus Y and La/Yb versus Yb

A relatively broad distribution is observed in La/Yb versus Yb (Fig. 2a) of the experimental adakites, a result of variable degrees of partial melting, at different P-T conditions, and different basaltic starting materials, but the overall fractionation of La from Yb coupled with the relative depletion in Yb is obvious. Similar to other adakites, associated with subduction, O-type adakites from Chinashow variable but somewhat less pronounced fractionation of La from Yb, with comparable depletion in Yb. In contrast, C-type adakites show more highly variable, and in some cases extreme, degrees of REE fractionation and Yb depletion (comparable to the range seen in the experimental melts). Intermediate and felsic samples (tonalites and sodic ignimbrites) from the Cordillera Blanca are distributed over a range comparable to the experimental melts and C-type adakites.

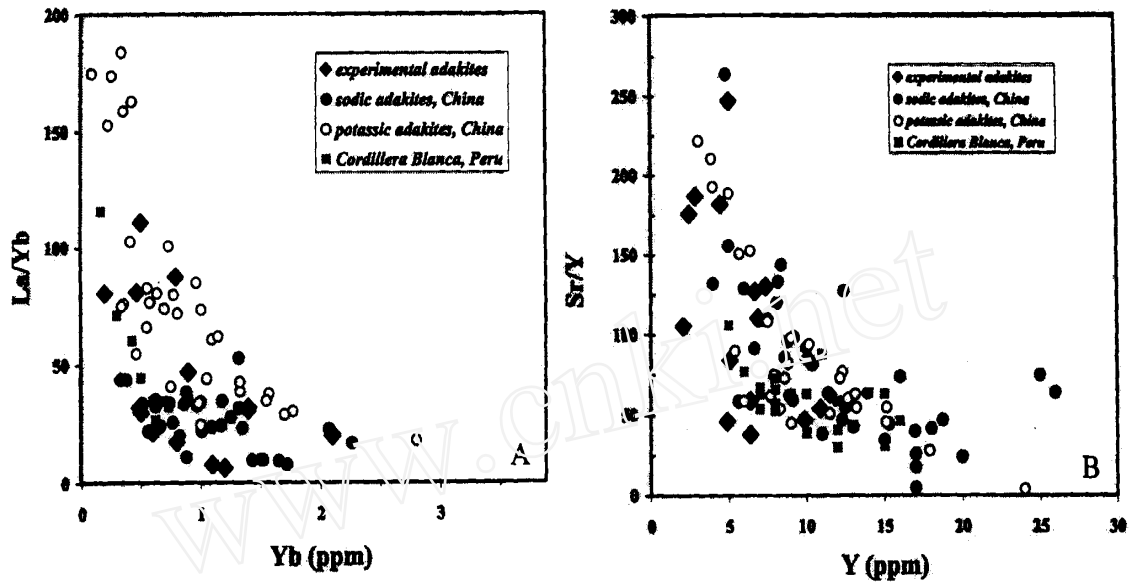


Fig. 2 La/Yb versus Yb (a) and Sr/Y versus Y (b) for: experimental melts at 1- 4 GPa (samples of Rapp and Watson (1995) and Rapp *et al.* (1999); "normal" sodic adakites from eastern and central China (data of Li *et al.*, 2001; Wang and Zhang, 2001; Xiaong *et al.*, 2001; Xu *et al.*, 2000, 2001; Zhang *et al.*, 2001), potassic adakites from eastern China (data of Pan *et al.*, 2001; Wang and Zhang, 2001; Wang *et al.*, 2001; Zhang *et al.*, 2001), and tonalites and quartz diorites, and leucogranodiorites from the Cordillera Blanca batholith (squares with cross, data of Atherton and Petford, 1993; Petford and Atherton, 1994)

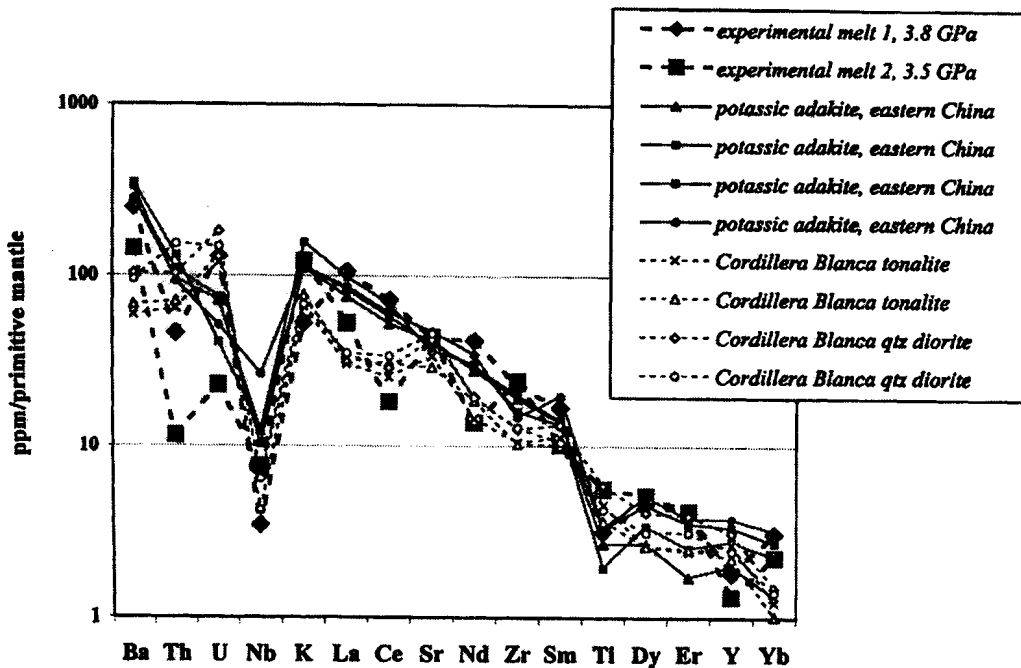


Fig. 3 Trace-element abundance patterns, normalized to primitive mantle (Sun and McDonough, 1989) for experimental adakite melts of basalt # 1 at 3.8 and # 2 at 3.5 GPa (Rapp and Watson, 1995; Rapp *et al.*, 1999), coexisting with rutile-bearing eclogite residues. Similar patterns are observed in adakite melts from lower pressure experiments at 1- 3 GPa (data of Table 2 and unpublished data of Rapp and Shimizu). Shown for comparison are the patterns for potassium-rich adakites from eastern China (data of Wang *et al.*, 2001; samples DBS 1019 and DBS JZ05); Pan *et al.*, 2001; sample 2174-1); Zhang *et al.*, 2001; sample 134a), and tonalites and quartz diorites from the Cordillera Blanca batholith, NW Peru (Petford and Atherton, 1996)

No clear distinctions are apparent between these data sets in terms of Sr/Y versus Y (Fig 2b), with adakites from melting experiments, eastern China and the Peruvian Andes all showing the same general distribution trend, that of increasing fractionation of Sr from Y, coupled with decreasing Y concentrations. Based on the experimental samples, the range in the distribution of the data for the natural adakites is attributable to a combination of factors, including variable degrees of melting, at various P-T conditions, of compositionally variable yet broadly basaltic sources.

#### Mantle-normalized trace element abundance patterns

The overall distribution of trace elements during partial melting is controlled by mineral/melt partition coefficients for adakite liquids and crystalline phases present in the residue of melting (dominantly garnet, clinopyroxene, and rutile at pressures above 2 GPa). The characteristic trace element signature of adakite is distinguished by an overall enrichment in large ion-lithophile elements (including Ba, Th, U, K, LREEs, Sr), and strong depletions in HFSEs (*e.g.*, Nb, Ti) and the HREEs and Y (Defant and Drummond, 1990; Kay *et al.*, 1993). These features are the consequence of equilibration between adakite melts and eclogitic residues containing rutile, typify adakite liquids in equilibrium with eclogitic residues at 1.2- 3.8 GPa, and are exemplified by the mantle-normalized trace-element abundance patterns (spidergrams) for two liquids from melting experiments at 3- 4 GPa (Fig 3). Patterns for C-type adakites from eastern China (data from Pan *et al.*, 2001; Wang and Zhang, 2001; Wang *et al.*, 2001) show close similarities to the experimental melts, with general enrichment in LLEs, sharply negative Nb-anomalies, strong depletions in HREEs and Y, and variable relative depletion in Ti. The Chinese adakites, however, are distinguished by LREE and LLE abundances that are as high or higher than liquids formed at 3.8 GPa from the most potassic basalt from Table 1; Ba, Th and K abundances are especially high in the Chinese samples. Potassic adakites have higher abundances of LREEs (La and Ce), MREE (Nd and Sm), Th and Ba, and lower U, when compared to tonalites and leucotonalites from the Cordillera Blanca. Similar comparisons with other Na-granitoids (*i.e.*, adakites) interpreted as lower crustal melts reveal similar compositional contrasts with the potassic adakites.

Nevertheless, the trace-element abundance patterns in Fig 3 do strongly suggest that all these samples are related in the sense that they originated by partial melting of hydrous, garnet-bearing metabasalt, and the adakite geochemical signature was imparted through mineral/melt partitioning with the residual phase assemblage (garnet + clinopyroxene).

## Discussion

Potassium-rich adakites from eastern China appear to be compositionally unique among granitoids possessing the adakite geochemical signature and interpreted as lower crustal melts. Similarities between the major- and trace-

element characteristics of adakites from eastern China and experimental melts clearly indicate an origin by dehydration melting of garnet-bearing metabasalt. The metaluminous A/CNK ratios (~ 1.0- 1.1) and uniformly low Mg# 's (less than 40- 45) of the Chinese adakites suggest the following, respectively: (1) amphibole was absent or a minor phase in the residue of melting, and thus temperatures approached or exceeded the amphibole-out phase boundary (~ 1000 °C at 2 GPa), and (2) the parental adakite magmas had limited interaction with mantle (peridotitic) assemblages. At pressures above 1.5- 2.0 GPa, sphene and/or rutile play a crucial role in effecting the sharp negative anomalies in Nb, Ti and other HFSEs apparent in Figure 3.

Liquids produced at pressures above 4 GPa (Table 2 and Fig 1) have very low Na<sub>2</sub>O/K<sub>2</sub>O ratios (~ 0.5), but their Al<sub>2</sub>O<sub>3</sub> contents are much lower than those of the potassic adakites, and implied crustal thicknesses are unrealistic (> 150 km). Potassium-rich granitoids formed at slightly lower pressure (3.5- 3.8 GPa; see Table 2) correspond more closely to C-type adakites in terms of Na<sub>2</sub>O/K<sub>2</sub>O ratios and Al<sub>2</sub>O<sub>3</sub> contents, but would still require crustal thicknesses in excess of 120 km. Very low degrees of melting at pressures of 1- 3.5 GPa (40- 120 km depth) can produce high-potassium liquids, but these liquids possess A/CNK ratios that are too high (< 1.1- 1.2) and Al<sub>2</sub>O<sub>3</sub> concentrations that are too low to correspond to most adakites. Furthermore, such melts would also be expected to have the highest La/Yb and Sr/Y ratios, due to the partitioning effects of residual eclogitic phases, yet there is no obvious correlation between La/Yb or Sr/Y on the one hand, and Na<sub>2</sub>O/K<sub>2</sub>O ratios on the other. And adakites with low Na<sub>2</sub>O/K<sub>2</sub>O ratios occur over a range of SiO<sub>2</sub> contents, rather than at uniformly high SiO<sub>2</sub>, which would be expected for very low-degree partial melts.

More likely, the compositional distinctiveness of potassic adakites is either attributable to a basaltic source that is unusually enriched in potassium and other LLEs, or to the effects of assimilation of intermediate lithologies in the lower-middle crust. Isotopic studies clearly indicate some level of crustal involvement in eastern China, but adakites from the Andean Austral Volcanic Zone which show isotopic evidence for significant interaction with the sub-arc crust (*e.g.*, Burney and Reclus volcanoes; Killian and Stern, 1996) still have "normal", adakitic, Na<sub>2</sub>O/K<sub>2</sub>O ratios (3- 5), and significantly lower LLE abundances than the potassic adakites of eastern China.

These considerations lead us to believe that the geochemical peculiarities that distinguish potassium-rich adakites arise from some unique aspect of either their source, or the lower crustal material with which they interacted. Potassium-rich shoshonites with high LLE abundances (*e.g.*, Ba, Th, U, LREEs), erupted in eastern China in the early Cretaceous (Wang and Zhang, 2001) attest to mantle activity and suggest a possible "enriched" source for the C-type adakites that followed upon magmatic underplating and crustal overthickening. Alternatively, assimilation of compositionally intermediate granitic lithologies in the middle to lower crust (Taylor and

Table 2 Major element compositions of near-solidus melts from 3.8 to 5.0 GPa

| P (GPa)                            | 1.8   | 3.2   | 3.8   | 3.8   | 3.8   | 3.8   | 5.0   | 5.0   |
|------------------------------------|-------|-------|-------|-------|-------|-------|-------|-------|
| Temp (°C)                          | 1000  | 1100  | 1000  | 1100  | 1200  | 1000  | 1100  | 1200  |
| Oxide (wt%)                        |       |       |       |       |       |       |       |       |
| SiO <sub>2</sub>                   | 67.88 | 65.93 | 71.10 | 67.52 | 66.61 | 73.15 | 67.20 | 69.29 |
| TiO <sub>2</sub>                   | 0.49  | 1.31  | 0.79  | 1.72  | 1.88  | 0.68  | 1.78  | 2.19  |
| Al <sub>2</sub> O <sub>3</sub>     | 18.01 | 17.55 | 14.43 | 17.60 | 13.84 | 14.68 | 12.01 | 11.58 |
| FeO*                               | 2.27  | 3.15  | 1.16  | 3.45  | 3.68  | 1.33  | 4.02  | 4.13  |
| MnO                                | 0.24  | 0.12  | 0.02  | 0.07  | 0.04  | 0.03  | 0.05  | 0.06  |
| MgO                                | 1.49  | 1.02  | 0.43  | 1.53  | 1.48  | 0.67  | 1.65  | 1.84  |
| CaO                                | 1.25  | 1.47  | 1.36  | 2.55  | 2.42  | 2.98  | 3.37  | 2.56  |
| Na <sub>2</sub> O                  | 6.47  | 6.72  | 3.12  | 5.12  | 5.98  | 4.53  | 2.77  | 2.69  |
| K <sub>2</sub> O                   | 2.62  | 2.55  | 7.03  | 4.13  | 3.66  | 1.59  | 6.46  | 5.66  |
| P <sub>2</sub> O <sub>5</sub>      | n.a.  | n.a.  | 0.57  | n.a.  | 0.42  | 0.35  | n.a.  | n.a.  |
| Mg <sup>#</sup>                    | 0.28  | 0.37  | 0.40  | 0.44  | 0.42  | 0.47  | 0.42  | 0.44  |
| Al/CNK                             | 1.14  | 1.06  | 0.95  | 1.00  | 0.76  | 1.01  | 0.68  | 0.76  |
| Na <sub>2</sub> O/K <sub>2</sub> O | 2.47  | 2.64  | 0.44  | 1.24  | 1.63  | 2.85  | 0.43  | 0.48  |
| Melt (%)                           | 20    | 30    | 12    | 17    | 36    |       | 8     | 13    |
| Trace elements                     |       |       |       |       |       |       |       |       |
| La                                 | 8.8   | 15    | 20    | 16.6  | -     | 72    | 39.4  | -     |
| Yb                                 | 0.6   | 0.5   | 0.6   | 0.9   | -     | 1.5   | 0.7   | -     |
| La/Yb                              | 15    | 30    | 36    | 18    | -     | 48    | 56    | -     |
| Sr                                 | 390   | 820   | -     | 854   | -     | 907   | 1663  | -     |
| Y                                  | 4.5   | 4.5   | -     | 10    | -     | 8.2   | 13.5  | -     |
| Sr/Y                               | 87    | 182   | -     | 85    | -     | 111   | 123   | -     |

McClenan, 1995) could explain the preferential enrichment in potassium and other LLEs. In either case, C-type adakites are considered to be the products of partial melting at the base of magmatically thickened lower crust; adakite magmatism was followed by a period of crustal extension and thinning, possibly in response to the foundering of dense eclogitic residues of melting, lower crustal delamination, and inflow of hot asthenospheric mantle (Wang and Zhang, 2001).

"Flat slab subduction" models based on seismic, geochemical and geochronologic evidence have recently been proposed for the Andes in Peru and southern Ecuador (Beate *et al.*, 2001; Gutscher *et al.*, 1999; Gutscher *et al.*, 2000), offering an alternative to the lower crustal melting models of Atherton and Petford (1993). In these models, adakite magmatism is the result not of partial melting in the lower crust, but a consequence of slab melting. Subduction of the thick and buoyant "lost Inca Plateau" proceeded at a shallow angle and flattened out after penetrating 300-400 km inboard of the trench (Beate *et al.*, 2001), providing an alternative (melting of the flat slab) to the lower crustal melting models for adakite volcanism in northern Peru and southern Ecuador. It remains to be seen whether or not such geodynamic models are applicable to eastern China. Kimura *et al.* (1990) suggest that eastern Asia was being intensely deformed in the early Cretaceous as a consequence of collision between the Indochina Block and the Eurasian continent, arguing that these events controlled the evolution of subduction along the eastern continental margin. Yanshanian adakite magmatism in eastern China would then

have to be understood within the context of this more complex geodynamic setting, and any petrogenetic model for potassium-rich adakites would have to account for their compositional uniqueness.

## Conclusions

Potassium-rich adakites of eastern China appear to be compositionally among adakites formed by either "slab melting" or "lower crustal melting" of broadly basaltic sources. The distinctiveness of the potassic adakites is attributable either to unique aspects of the source (alkalic basalt enriched in LLEs), or to the nature of the material with which it reacts during transport and emplacement. For most of the eastern China adakites, there is evidence for only limited mantle involvement in their petrogenesis, given their generally low Mg<sup>#</sup>'s (< 45); assimilation of granulitic lithologies in the lower to middle crust could explain both the low Mg<sup>#</sup>'s and the preferential enrichment in potassium and other LLEs, while being consistent with the isotopic evidence. Clearly, additional experimental study (including melting and assimilation experiments) is needed to better constrain the petrogenesis of potassium-rich adakites, and their role in crustal growth in eastern China, but their inclusion in the adakite family of granitoids, with origins in the dehydration melting of garnet-bearing metabasalt, seems appropriate.



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